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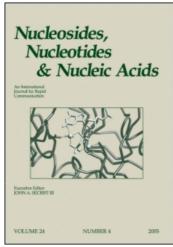
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Syntheses of Triazole and Pyrimidine Homo-C-nucleosides

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SYNTHESES OF TRIAZOLE AND PYRIMIDINE HOMO-C-NUCLEOSIDES

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Abstract: The preparation of 2,3:5,6-di-O-isopropylidene- α , β -D-mannofuranosyl-pyrimidine and 2,3-O-isopropylidene-5-O-trityl- α , β -D-ribofuranosyl-pyrimidine and triazole homo-C-nucleosides 3, 4, 6, 10-17 by different synthetic approaches is discussed and their anomeric configurational assignments are made based upon 1H- and 13C-NMR data.

Since Holy1 reported the interesting biological activity of homonucleotides, several syntheses of homo-C-nucleosides have been reported. Thus, Gensler2,3 reported several syntheses of triazole homo-C-nucleosides, and Just4 the syntheses of 6-azauracil and pyrazole homo-C-nucleosides. Secrist5 has prepared some 6-substituted 4- pyrimidinones, and Sato some pyrimidine homo-C-nucleosides6, homoshowdo-mycin, homopyrazomycin7, and pyrazole homo-C-nucleosides8,9. Cupps10 and Katagiril1 have prepared some pyrimidine homo-C-nucleosides, some of them as intermediate products in the synthesis of a pyrrolo [3,2-d]pyrimidine nucleoside12.

As a part of our work on the synthesis of antibiotic C-nucleosides and analogs13,14,15, we show in this paper the synthesis of some homo-C-nucleosides and the transformation of several different types of C-glycosides into homo-C-nucleosides potentially interesting for the preparation of C-nucleosides.

Troportion
$$CH_2COCH_2CO_2Me$$
 $CH_2COCH_2CO_2Me$ $CH_2COCH_2CO_2Me$ $CH_2COCH_2CO_2Me$ $CH_2COCH_2CO_2Me$ $CH_2COCH_2CO_2Me$ $CH_2COCH_2CO_2Me$ $CH_2COCH_2COCH_2CO_2Me$ $CH_2COCH_2COCH_2CO_2Me$ CH_2COCH

In a first approach we employed the α,β -mixture of me-4-D-ribo- or 4-D-manno-furanosyl-2-oxo-butanoate derivatives ($1\alpha\beta$ and $2\alpha\beta$ respectively), two products previously reported by us13, and with a synthetically interesting β -ketoester group. Thus, the reaction of $1\alpha\beta$, 2α and 2β separately with S-methyl thiourea hydroiodide in methanol /water and in the presence of K2CO3 gave the corresponding 4-hydroxy-2-methylthio-6-(D-ribo or D-manno-furanosylmethyl) pyrimidine derivatives $(3\alpha\beta, 4\alpha \text{ and } 4\beta)$. The mixture $3\alpha\beta$ was partially resolved by thick layer chromatography. The anomeric configuration of these products was determined by and 13C-NMR spectroscopy. Thus, the J3',4' values found for 3α (<1 Hz) and for 3β (4 Hz) are in agreement with the previously reported data13a for α -D-ribofuranosyl- and β -Dribofuranosyl-C-glycosides and C-nucleosides, respectively. Similarly the J1',2' values found for 4α (<1 Hz) and 4β (3.6 Hz), are also in agreement with the assigned configuration of α -D-mannofuranosyl- and β -D-mannofuranosyl-C-glycosides respectively13a. These assignments were confirmed by the 13C chemical shifts of 2',3'-0-isopropylidene groups. Thus, the "exo" anomer or β -D-ribofuranosyl derivative 3β ,

larger chemical shift difference between the two methyls of the 2',3'-O-isopropylidene group (1.85 ppm), a higher chemical shift for the corresponding quaternary carbon CMe2 (114.07) and methylene group (41.97), than the "endo" or α -anomer 3α (1.22 ppm, 112.38 ppm and 38.09, respectively)16. Similarly, the "exo" or α -D-mannofuranosyl derivative 4α shows higher chemical shift values for the methylene group (38.77 ppm) than the "endo" or β -anomer 4β (36.44 ppm).

Alternatively, the reaction of $1\alpha\beta$ with benzylamine in benzene, with azeotropic elimination of water, yielded the corresponding enamines $5\alpha\beta$ in good yield which were easily converted into the triazole homo-C-nucleosides $6\alpha\beta$, also in good yield, by reaction with p-toluenesulfonyl azide in acetonitrile and in the presence of ethylamine. Both isomers were separated by column chromatography and characterized by analytical data. Thus, 1H-NMR shows a 33',4'<1 Hz for 6α and 4.1 Hz for 6β , and 13C-NMR shows a minor chemical shift difference between the two isopropylidene methyls for the "endo" or α -anomer 6α (1.58 ppm) than for the "exo" or β -anomer 6β (1.89 ppm).

Finally, the Wittig reaction between the partially protected sugars $2,3-\underline{0}$ -isopropylidene-D-ribofuranose (7) and 2,3:5,6-di- $\underline{0}$ -isopropylidene-D-mannofuranose (8) with (6-chloropyrimidin-4-yl)methylenetriphenylphosphorane (9), gave the D-ribofuranosyl or D-mannofuranosyl-methylpyrimidine-homo-C-nucleosides ($10\alpha\beta$)10 and ($14\alpha\beta$), respectively, besides minor quantities of the unsaturated products $10EZ^{10}$ and 14EZ. The α - and β -anomers 14α and 14β were separated by column chromatography.

In connection with our studies on the transformation of these and similar C-glycosides into C- and homo-C-nucleosides, we are interested in studying the reactivity of compounds ($10\alpha\beta$) and ($14\alpha\beta$) with different nucleophiles, (we have substituted the 4-chloro by OMe, OH and NH2) to modify the reactivity at C-5 against electrophiles. Thus, the

Tro
OH

(10 EZ)

(14 EZ)

(14 EZ)

(14 EZ)

(10 EZ)

(14 EZ)

(14 EZ)

(10 EZ)

(14 EZ)

(14 EZ)

(10 EZ)

(10 EZ)

(14 EZ)

(14 EZ)

(15
$$\alpha\beta$$
) X = Cl

(15 $\alpha\beta$) X = OMe

(15 $\alpha\beta$) X = OMe

(15 $\alpha\beta$) X = OMe

(16 $\alpha\beta$) X = NH₂

(17 $\alpha\beta$) X = OH

reaction of $10\alpha,\beta$ separately with methanolic ammonia give the 4-methoxy derivatives $11\alpha,\beta$ besides the 4-amino analogues $12\alpha,\beta$. Reaction of 10 with aqueous potassium hydroxide, led to the corresponding α,β mixture $13\alpha\beta$ which could be separated by thick layer chromatography. The same product was obtained from Ni-Raney reduction of $3\alpha\beta$. Similarly, treatment of 14 with methanolic ammonia gave $15\alpha\beta$ and $16\alpha\beta$, while with aqueous potassium hydroxide or reduction of 4β with Ni-Raney led to 17β .

The annomeric configurational assignments of all these compounds were determined by $^{1}\text{H-}$ and $^{13}\text{C-NMR}$ spectroscopy 13a,16 . Thus, as above, all α -D-ribo-derivatives showed smaller 13 ',4' <1 Hz, chemical shift differences between the isopropylidene methyls (1.19 to 1.22 ppm), and chemical shifts for quaternary carbon of isopropylidene group (112.32 to 112.38 ppm) and CH₂ (37.75 to 38.09 ppm), than β -D-ribo-anomers (3.0 to 4.0 Hz, 1.74 to 1.85, 114.07 to

114.26 and 41.53 to 41.97 ppm, respectively), (Tables I to VI). Reactivity study of these products with different nucleophiles is in progress. In all cases, the basic medium causes anomerization, yielding a greather proportion of the "endo" anomer, showing that these anomers are the thermodinamically more stable ones, in agreement with previously reported analogous observations 13,16.

EXPERIMENTAL RESULTS

Melting points are uncorrected. Infrared spectra were recorded with a Beckman Aculab IV spectrophotometer. 1H and 13C NMR spectra were recorded with a Bruker WP spectrometer (CDCl3 was used as solvent and internal reference). Mass spectra were obtained with a Hewlett 5988A, or with a Kratos MS25RFA. Elemental analyses were carried out in the Microanalysis Service of the University of Malaga, in a Perkin Elmer 240. High performance liquid chromatography (H.P.L.C.) was carried out on a Hewlett Packard 1084 B, using a 254 nm UV detector. Preparative thick layer chromatography was performed with silica gel 60 PF254 (Merck 7747), thin layer chromatography with silica 60 F254 (Merck 5719), column chromatography with silica gel 60 (Merck 7734).

Synthesis of $6-(2,3-Q-isopropylidene-5-Q-trityl-\alpha-$ and $\beta-D-ribofuranosyl)$ methyl-4-hydroxy-2-methylthiopyrimidine $(3\alpha\beta)$. To a solution of 4.46 g (8.41 mmol) of $1\alpha\beta^{13b}$ in 15 mL of ethanol were added 3.55 g (17.23 mmol) of S-methyl thiourea hydroiodide dissolved in 6.35 mL of water and 5.97 g of K2CO3 in 4.3 mL of water. The reaction was left at r.t. for 95 hrs, and then heated at reflux for one additional hour. The solution was neutralized with diluted hydrochloric acid, leading to a gummy precipitate which was purified by column chromatography on silica gel (hexane: ethyl acetate 3:1) to yield 0.98 g of unaltered $1\alpha\beta$, and 2.48 g of a mixture of

w.
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AB.
-

C-NMR C	13C-NMR chemical shifts (6) of D-ribo compounds 3, 6a, 11 and 12 (6, 50 MHz).	hifts (6) of D-r	ibo com	pounds 3,	, 6a, 11	and 12	(6, 50)					
punodu	Compound C-1' C-2' C-3' C-4'	c-5	ر- ج	-7-0	C-2	C-5' C-2 C-4 C-6 C-5	7- 5	9-5			CHe2	CMe2 Trity(Trityl Others
	()) 计自己有利证券经济自由的自己有价的分配的原理的自己的证明证据的证明的分别的分别	******	**********		********	***************************************	######################################	*******	222222	11 11 11 11 11 11	20022		
34	83.59	83.59-83.44-82.29-80.28	2,29-80.	28	65.38	165.88-1	165.24-1	61.29	109.18	26.40	25.18	112.38	65.38 165.88-165.24-161.29 109.18 26.40 25.18 112.38 87.28-143.60-128.62-127.87-127.13 38.09 (CH2) 12.89 (SMe)
3,8	84.65	84.65-83.47-82.55-82.33	2.55-82.	33	64.23	164.90-1	1-12-191	43.59	109.24	27.50	25.65	114.07	64.23 164.90-161.21-143.59 109.24 27.50 25.65 114.07 87.00-143.83-128.72-127.83-127.06 41.97 (CH2) 13.23 (SMe)
11a	83	83,45-82,18-80,47	8-80.47-		64.51	170.00-1	167.32-1	57.97	107.58	26.36	25.17	112.38	170.00-167.32-157.97 107.58 26.36 25.17 112.38 87.30-143.64-128.66-127.81-127.02 37.78 (CH2) 53.58 (OMe)
11,8	84.41	84.41-83.43-82.94-82.63	2.94-82.	63	64.17	169.92-1	168.52-1	57.97	107.70	27.52	25.68	114.26	169.92-168.52-157.97 107.70 27.52 25.68 114.26 87.10-143.82-128.66-127.81-127.03 41.60 (CH2) 53.58 (OMe)
12a	83	83.32-82.18-80.41	8-80.41-		64.22	166.32-1	162.97-1	58.28	104.38	26.34	25.12	112.32	64.22 166.32-162.97-158.28 104.38 26.34 25.12 112.32 87.06-143.57-128.57-127.82-127.05 37.75 (CH2)
12,8	84.25	84.25-83.44-82.84-82.01	2.84-82.		64.17	165,19-1	162.88-1	58.23	104.61	27.47	25.63	114.16	64.17 165.19-162.88-158.23 104.61 27.47 25.63 114.16 86.60-143.79-128.70-127.62-127.06 41.53 (CH2)
8488888	"我们我还有他们们在我们身份现代和他们的自己的自我们将我有我们你们们的我们然将我们的目标	**************************************		*******	***********	********	*********	*******	******	1	****	****	

a For compounds 6 α and 6 β see the experimental part.

TABLE 11

14-NMR chemical shifts of D-ribo products 3, 11, 12 and 13a (6, 200 MHz)	nical shif	ts of [0-ribo pr	oducts 3,	. 11, 12	and 13a	(8, 200 ₺	(Hz)							
Compound	H-1	H-2+	H-3	17-H	H-5'a	H-5.P	H-2: H-3: H-4: H-5:a H-5:bMe2C Tr	• • • • • •	٦٢	н-2 н-5		CH2 Others		Others	
	***********	*******			********		111111111111111111111111111111111111111				*******	## ## ## ## ## ## ## ## ## ##)) 	1) 1) 1) 1) 1)	## ## ## ## ## ## ## ## ## ## ## ## ##
3@	5	.00-4.	III 02	4.18 t	3.35 de	13.08 dc	11.55 s	1.32 s	7.5-7.2 ш	;	6.22 s	5.00-4.70 m 4.18 t 3.35 dd 3.08 dd 1.55 s 1.32 s 7.5-7.2 m 6.22 s 2.90 d 2.95 d 2.18 s (SMe)	55 d	2.18 s	(SMe)
3,8	4.35 m	4.55	3d 4.65 d	d 4.12 c	3.25 de	13.15 dc	11.54 s	1.30 s	7.5-7.2 m	;	6.10 s	4.55 dd 4.65 dd 4.12 c 3.25 dd 3.15 dd 1.54 s 1.30 s 7.5-7.2 m 6.10 s 2.80 d	:	2.52 s (SMe)	(SMe)
110	4.60 m	4.77	3d 4.72 d	d 4.19 t	3.20 dc	13.06 dc	11.52 s	1.31 s	7.5-7.2 m	8.72 d	6.76 d	4.77 dd 4.72 dd 4.19 t 3.20 dd 3.06 dd 1.52 s 1.31 s 7.5-7.2 m 8.72 d 6.76 d 3.07 d	:	3.96 s (OMe)	(OMe)
118	4.34 dd	4.57	3d 4.61 d	d 4.12 de	13.26 do	13.13 dc	11.51 s	1.31 s	7.5-7.2 ₪	8.72 d	6.68 d	4.57 dd 4.61 dd 4.12 dd 3.26 dd 3.13 dd 1.51 s 1.31 s 7.5-7.2 m 8.72 d 6.68 d 3.11 dd 2.94 dd 3.92 s (OMe)	% dd	3.92 s	(OMe)
12a	4.60 m	4.74	3d 4.70 b	d 4.20 bt	3.18 do	13.07 dc	1 1.51 s	1.33 s	7.5-7.2 m	8.49 d	6.43 d	4.74 dd 4.70 bd 4.20 bt 3.18 dd 3.07 dd 1.51 s 1.33 s 7.5-7.2 m 8.49 d 6.43 d 2.97 m 5.00 bs (NH2)	:	5.00 bs	(NH2)
128	4.35 m	4.55	5-4.65 m-	- 4.13 m	3.27 do	13.12 dc	11.50 s	1.30 s	7.5-7.2 m	8.49 d	6.33 d	4.55-4.65 m 4.13 m 3.27 dd 3.12 dd 1.50 s 1.30 s 7.5-7.2 m 8.49 d 6.33 d 3.95 dd 3.85 dd 4.92 bs (NHZ)	35 dd	4.92 bs	(NH2)
13α	7	.80-4.6	ш 05	4.20 t	3.24 do	13.10 de	1 1.52 s	1.32 s	7.4-7.1 m	8.10 s	6.50 s	80-4.60 m 4.20 t 3.24 dd 3.10 dd 1.52 s 1.32 s 7.4-7.1 m 8.10 s 6.50 s ·· 2.96 d ·-	•		

a For compounds 5α , 5β , 6α and 6β see the experimental part.

4.30 m 4.53 dd 4.63 dd 4.13 dd 3.27 dd 3.13 dd 1.52 s 1.31 s 7.5-7.2 m

8.03 s 6.40 s 2.95 dd 2.82 dd

TH-WMP COLD INC TABLE 111

Compound	CH2a-H1'	CH2b-H1	CH28-H1' CH25-H1' CH28-CH2b H1'-H2' H2'-H3'	H11-H2	H21-H31	H31-H41		H41-H51P	H4'-H5'a H4'-H5'b H5'a-H5'b	H2-H5
## ## ## ## ## ## ##				******	**************************************			## ## ## ## ## ##	***************************************	
3α	;	;	÷	;	0.9	⊽	5.0	5.0	12	,
3β	2	-	;	5.0	6.0	4.0	5.0	2.0	10	,
110	,	:	:	3.5	6.0	-	4.5	;	9.5	1.0
11,8	5.0	8.0	14	4.0	6.5	3.0	7.0	4.5	10	1.0
12a	;	÷	:	3.2	6.0	~	4.5	4.5	10	1.5
12,8	;	•	1	:	:	;	3.5	4.5	10	1.0
13a	,	:	;	3.6	5.9	0.5	4.1	4.5	10	1.0
13,8	5.5	ఱ	14	4.5	6.5	3.5	0.4	4.1	10	1.0

a For compounds 5α , 5β , 6α and 6β see the experimental part.

..... CMe2 C-5 9-J 13c-NMR chemical shifts of D-manno compounds 4, 14, 15, 16 and 17 (6, 50 MHz). Compound CH2 C-1 C-2 C-3 C-4 C-5 C-6 C-2 C-6 C-6

Compound CH2 C-1' C-2'	CH2	-1-3	i	C-3 C	-4- C	-51	c-6,	C-3' C-4' C-5' C-6' C-2 C-4 C-6	7-0	9-0	c-5	χ̈́	- CMe2	:	ÇMe2	Others
4α	38.77	82	.81-82.8	11-80.86-	73.	.37 6	7.03	38.7782.81-82.81-80.86 73.37 67.03 164.43-164.23-161.55	54.23-16		109.22	24.78-25.20-26.16-26.94	-26.16-26	5.94	112.87	14.50 (SMe)
84	36.44	81.63	-81.21-8	30.73-79.	59 73.	9 60.	6.84	36.4481.63-81.21-80.73-79.59 73.09 66.84 165.32-165.17-161.17	55.17-16	1.17	109.00	24.61-25.24-25.74-26.85	-25.74-26	5.85	112.47	13.16 (SMe)
14,8	36.48	81.67	-81.16-8	30.71-79.	70 73.	9 90.	. 29.9	36.4881.67-81.16-80.71-79.70 73.06 66.67 158.57 169.20-161.09	59.20-16		121.73	26.77-25.68-25.24-24.52	-25.24-24		112,60-108.99	
15β	36.24	81.75	-81,35-8	30.82-80.	13 73.	.15 6	6.89	36.2481.75-81.35-80.82-80.13 73.15 66.89 157.61167.06	167.06	:	107.75	26.85-25.79	-25.32-24	67	26.85-25.79-25.32-24.67 112.54-109.03	53.78 (OMe)
16,8	36.24	81,48	-81.22-8	30.62-80.	06 73.	.03	69.9	36.2481.48-81.22-80.62-80.06 73.03 66.69 158.07 165.43-162.98	55.43-16	2.98	104.46	26.74-25.68	-25.15-24	52	26.74-25.68-25.15-24.52 112.31-108.91	
17,8	36.54	81.75	-81.32-8	30.79-79.	54 73.	.08	68.9	36.5481.75-81.32-80.79-79.54 73.08 66.89 147.95 166.42-164.20	56.42-16	4.20	112.61	26.88-25.77	-25.28-24	99.	26.88-25.77-25.28-24.66 114.35-109.06	
					1											

TABLE V

1H-NMR chemical shifts of D-manno products 4, 14, 15, 16 and 17 (6, 200 MHz)

Commoning CHO, CHO, H-1: H-2: H-4: H-5: H-6:a

Compound	СИ2а	сн2р	H-11	H-2,	н-3	H-2: H-3: H-4:	н-51	H-6'a H-6'b	q.9-	1 1	Me2C	Me2C	н-5	Others	
			11 11 11 11 11 11 11 11	# # # # # # # # # # # # # # # # # # #	***		**********	10 11 11 11 11 11 11 11 11 11 11 11 11 1	11 11 11 11 11 11 11 11 11 11 11 11 11	************	() 	*********	54 11 11 11 11 11 11 11 11 11 11 11		*******
40	2.63	E S	4.50 t	P 69.7	4.82 dd	13.81 dd	4.37 ddd	4.07 dd 3.	. pp 26	1.49, 1	.45, 1.	37, 1.33	4s 6.08	2.63 m 4.50 t 4.69 d 4.82 dd 3.81 dd 4.37 ddd 4.07 dd 3.97 dd 1.49, 1.45, 1.37, 1.33 4s 6.08 s 2.50 s (SMe) 13 bs (OH)	3 bs (OH)
θ †	2.97 dd 2.84		3.95 m	4.65 dd	1 4.76 dd	13.50 dd	4.39 ddd	7.06 d 4.1	. p sc	1.50, 1	.43, 1.	37, 1.33	4s 6.17 :	dd 3.95 m 4.65 dd 4.76 dd 3.50 dd 4.39 ddd 4.06 d 4.05 d 1.50, 1.43, 1.37, 1.33 4s 6.17 s 2.56 s (SMe)	
140	2.88 dd 2.96		4.50 bt	4.70 dd	14.84 dd	3.89 dd	4.41 ddd	dd 4.50 bt 4.70 dd 4.84 dd 3.89 dd 4.41 ddd 4.07 dd 3.96 dd 1.51, 1.46, 1.39, 1.34 4s 7.30 d 8.92 d	. pp 96	1.51, 1	.46, 1.	39, 1.34	4s 7.30 c	d 8.92 d	
14,8	3.14	P 1	3.95 m	4.67 dd	i 4.75 dd	13.53 dd	4.38 ddd	3.14 d 3.95 m 4.67 dd 4.75 dd 3.53 dd 4.38 ddd - 3.95-4.15 m - 1.51, 1.42, 1.37, 1.34 4s 7.36 d 8.90 d	E	1.51, 1	.42, 1.	37, 1.34	4s 7.36 c	d 8.90 d	
15α	2.85 dd 2.76	2.76 dd	4.45 bt	4.69 bd	1 4.80 dd	13.80 dd	4.37 ddd	4.03 dd 3.	. pp %	1.44, 1	.40, 1.	33, 1.30	09.9 st	dd 4.45 bt 4.69 bd 4.80 dd 3.80 dd 4.37 ddd 4.03 dd 3.94 dd 1.44, 1.40, 1.33, 1.30 4s 6.60 d 8.68 d 3.94 s (OMe)	(OMe)
15 <i>β</i>	3.01 dd 2.93		3.90 ш	4.57 dd	1-4.65 dd	3.40 dd	dd 3.90 m 4.57 dd-4.65 dd 3.40 dd 4.30 m	3.95 dd 3.	. pp	1.41, 1	.32, 1.	27, 1.24	4s 6.59 c	3.95 dd 3.91 dd 1.41, 1.32, 1.27, 1.24 4s 6.59 d 8.59 d 3.86 s (OMe)	(OMe)
16α	2.76 dd 2.67		pg	4.71 d	4.80 dd	4.71 d 4.80 dd 3.81 dd	;	:	•	1.48, 1	.43, 1.	34, 1.31	4s 6.31 (1.48, 1.43, 1.34, 1.31 4s 6.31 d 8.47 d 4.90 bs (NH2)	(NH2)
168	2.99 dd 2.90	2.90 dd	dd 3.95 m		1-4.75 dd	3.47 dd	4.37 ddd	4.65 dd-4.75 dd 3.47 dd 4.37 ddd 4.05 dd 3.99 dd	Pb 64	1.48, 1	.40, 1	34, 1.31	4s 6.38 c	1.48, 1.40, 1.34, 1.31 4s 6.38 d 8.47 d 4.90 bs (NH2)	(NH2)
17a	2.67 m -	E /	;	4.66 dd	1 4.80 dd	4.66 dd 4.80 dd 3.78 dd	;	;		1.46, 1	.42, 1.	33, 1.30	4s 6.32	1.46, 1.42, 1.33, 1.30 4s 6.32 s 8.08 s (H-2)	
17,8	2.91	:==	3.91 dt	4.63 dd	4.73 dd	3.46 dd	4.36 ddd	4.04 dd 3.9	. pp 86	1.47, 1	.39, 1.	33, 1.30	48 6.40	2.91 m 3.91 dt 4.63 dd 4.73 dd 3.46 dd 4.36 ddd 4.04 dd 3.98 dd 1.47, 1.39, 1.33, 1.30 4s 6.40 s 8.08 s (H-2)	
					1 1 1 1	1			1 1 1	1			1 1 1 1 1 1 1 1 1		

H41-H51 H51-H618 H51-H616 H618-H616 H2-H5 1H-NMR coupling constants of D-manno products 4, 14, 15, 16 and 17 Compound CH2a-H1 CH2b-H1 CH2a-CH2b H11-H2: H2:-H3: H3:-H4: TABLE VI

				1]			2	2	5	1
ν,	7.5	7.5		<1	6.0	3.6	7.7	7.7 6.3	A 11 11 11	8.7	# (f H :
84	4.9	4.9	16	3.6	6.0	3.6	7.4	5.7	5.0	;	;
14a	6	80	14	0.5	6.0	7.0	7.2	6.0	4.0	8.0	1.5
14,8	9.9	9.9	;	3.5	6.1	3.5	7.2	8.7	5.0	0.6	1.5
15a	7.5	7.5	14.5	<0.5	6.0	3.5	7.5	6.0	4.5	8.0	1.5
15.8	7	6.2	14.5	3.5	0.9	3.5	7.5	6.0	4.0	8.0	1.5
16a	7.5	7.5	13	⊽	6.0	3.5	7.0	;	÷	;	1.0
16,8	9	7.2	4	3.5	6.0	3.5	7.0	6.0	5.0	8.5	1.0
17α	:	D 0 0	:	0.7	, 0	3.5	7.5	;	:	:	1.5
17,8	i	;	:	3.6	6.0	3.5	7.6	o;	3.8	8.8	1.5

epimers $3\alpha\beta$ (66.48% yield), which could be partially separated by thick layer chromatography (hexane:ethyl acetate 5:1). Rf: 0.85 (3α) and 0.90 (3β) (hexane:ethyl acetate 1:3); UV λ max (MeOH): 285 nm (ϵ 7.8x10³) and 227 nm (ϵ 1.6x10⁴); IR ν max (KBr): 3500-3300, 3100-3000, 2980, 2930, 2860, 1650, 1575, 1530, 1445, 1380, 1210, 1155, 1100, 1070, 860, 760, 740, 700 cm-1.

Anal. Calc. for C33H34N2O5S: C, 69.45; H, 6.00; N, 4.90. Found: C, 69.51; H, 6.12; N, 5.02.

Synthesis of 6-(2,3:5,6-di-0-isopropylidene-α-D-mannofuranosyl)methyl-4-hydroxy-2-methylthiopyrimidine (4α). To a solution of 0.75 g (2.09 mmol) of methyl 4-(2,3:5,6-di-0-isopropylidene-D-mannofuranosyl)-3-oxobutanoate (2α)13a in 4 mL of ethanol were added 0.88 g (2.45 mmol) of S-methyl thiourea hydroiodide and 1.5 g of K2CO3 dissolved in 4 mL of water. The reaction was left at r.t. for 36 hrs and then heated at reflux for 1 h. The solution was neutralized with diluted acetic acid, concentrated and purified by column chromatography on silica gel (hexane:ethyl acetate 3:2) to yield 0.525 g (63 %) of 4α . Rf 0.14 (hexane:ethyl acetate 1:1); $[\alpha]_{D}^{20}$ +27° (c 0.1, MeOH); m.p. 124°C; UV λ max (MeOH): 288 nm (ϵ 1.1x104) and 238 nm (ϵ 0.99x104); IR ν max (KBr): 3600-3300, 2980, 2940, 2880, 1660, 1580, 1545, 1465, 1385, 1215, 1170, 1120, 1070, 850 cm-1.

Anal. Calc. for C₁₈H₂₆N₂O₆S: C, 54.25; H, 6.57, N, 7.03. Found: C, 54.28; H, 6.62; N, 7.11.

Synthesis of $6-(2,3:5,6-di-\underline{O}-isopropylidene-\beta-D-manno-furanosyl)$ methyl-4-hydroxy-2-methylthiopyrimidine(4 β). A solution of 0.56 g (1.56 mmol) of 2β 13a in 1 mL of ethanol, was mixed with a solution of 0.66 g (3.20 mmol) of S-methyl thiourea hydroiodide and 1.13 g of K2CO3 in 3.5 mL of water. Working as above, were obtained 0.419 g (67 %) of 4β . Rf 0.36 (hexane: ethyl acetate 1:2); $[\alpha]_D^{2O}$ -8.3° (c 0.79, MeOH); m.p. 68°C; UV λ max (MeOH): 282 nm (ϵ 1.35x104) and

232 nm (ϵ 1.45 x104); IR ν_{max} (KBr): 3600-3300, 2980, 2930, 2860, 1660, 1580, 1530, 1455, 1375, 1210, 1160, 1120, 1060, 840 cm-1.

Anal. Calc. for C₁₈H₂₆N₂O₆S: C, 54.25; H, 6.57, N, 7.03. Found: C, 54.31; H, 6.59; N, 7.11.

Synthesis of methyl 4-(2,3-Q-isopropylidene-5-Q-tri $tyl-\alpha$ and β -D-ribofuranosyl)3-benzylamino-2-butenoate (5 $\alpha\beta$). A mixture of 2.02 g (3.80 mmol) of $1\alpha\beta$ 13b and 0.41 g (3.83 of benzylamine in 15 mL of dry benzene were heated at reflux for 1h. Water was removed azeotropically by llation and the rest of the solvent was eliminated under The residue was chromatographed on silica gel (hexane-ethyl acetate 5:1) yielding 1.63 g (69.43 %) of a mixture of epimers $5\alpha\beta$. Rf 0.49 (hexane-ethyl acetate 3:1); $[\alpha]_{-}^{20}$ (c 1, CHCl3); UV h_{max} 268 nm (ϵ 3.84x104) and 256 nm $(\epsilon \ 3.16 \times 10^4)$; IR v_{max} (KBr): 3600-3200, 3100-3000, 2940, 1690, 1660, 1610, 1450, 1380, 860, 760, 740 and 700 1H-N.M.R. (δ): 9.1-8.9 (m, 2H, NH), 7.4-7.1 (m, 2 OH aromatics), 4.8-4.3 (m, 5H, H-1', H-2', H-3' of β , and H-2', H-3' of α -epimer), 4.71 (s, 1H, =CH β), 4.69 (s, 1H, =CH α), 4.48 (m, 2H, -CH2-Ph of β), 4.42 (m, 2H, -CH₂-Ph of α), 4.25-4.00 (m, 3H, H-1', H-4' of α and H-4' of β), 3.64 (s, 3H, MeO), 3.30 (dd, 1H, 4.4 and 10.1 Hz, H-5'a of α), 3.28 (dd, 1H, 4.3 and 9.9 Hz, H-5'a of β), 3.15 (dd, 1H, 4.4 and 10.1, H-5'b of α), 3.10 (dd, 1H, 4.3 and 9.9, H-5'b of β), 2.64 (d, 2H, 7.2 Hz, H-4a and H-4b of β), 2.55 (d, 2H, 6.3 Hz, H-4a and H-4b of α), 1.49, 1.43 and 1.31 (3s, 2x3H and 6H, Me₂C).

Anal. Calc. for C39H41NO6: C, 75.58, H, 6.66, N, 2.26. Found: C, 75.72; H, 6.80; N, 2.32.

Synthesis of 1-benzyl-4-methoxy-carbonyl-5-(2,3-Q-iso-propylidene-5-Q-trityl- α and β -D-ribofuranosyl)-1,2,3-tri-azole (6 $\alpha\beta$). To a solution of 1.45 g (2.34 mmol) of 5 $\alpha\beta$ and 0.24 g (2.37 mmol) of triethylamine in 3.7 mL of acetonitrile cooled at 0°C was added 0.48 g (2.43 mmol) of tosyl

azide. The reaction was left to rise to room temperature and after 48 h the solvent was evaporated under vacuo. The residue was dissolved in 30 mL of ethyl ether, washed with 0.18 g of NaOH in 20 mL of water, then with 0,1 g of NaOH in 10 mL of water and finally with water (2x10 mL), dried over Na₂SO₄ and chromatographed on silica gel (hexane-ethyl acetate 7:2), to yield the separated anomers 6α and 6β (0.59 and 0.40 g, 39% and 26.7% respectively).

6α: Rf: 0.24 (hexane-ethyl acetate 7:2); m.p. 52 °C; $[\alpha]^{20}$ +19.1° (c 1.01, CHCl₃); UV λ_{max} (CHCl₃): 290 nm (ϵ 545), 260 nm (ϵ 1540) and 241 nm (ϵ 6720); IR \mathbf{y}_{max} (KBr): 3090-3030, 2980-2870, 1745, 1720, 1570, 1450, 1375, 1250, 1210, 1075, 860, 760, 740 and 700 cm⁻¹. 1H-N.M.R. (δ): 7.5-7.0 (m, 2 OH, Ph and Tr); 5.82 and 5.67 (2d, 2x1H , 15.4 Hz, CH2-Ph), 4.9-4.6 (m, 3H, H-1', H-2', H-3'), 4.18-4.08 (bt, 3.3 Hz, 1H, H-4'), 3.67 (s, 3H, MeO), 3.37 (dd, 1H, 3.5 and 14.2 Hz, $CH_{2a}C=$), 3.22 (dd, 1H, 3.2 and 10.1 Hz, H-5'a), 3.09 (dd, 1H, 9.3 and 14.2 Hz, CH2bC=), 2.99 (dd, 1H, 3.44 and 10.1 Hz, H-5'b), 1.54 and 1.33 (2s, 6H, 13C-N.M.R. (δ): 161.89 (C=O), 143.49-127.07 (Ph and Tr), 112 (Me₂C), 83.66, 83.63, 82.03 and 81.14 (C-4', C-3', C-2' and C-1'), 65.32 (C-5'), 52.15 (CH₂-C=), 51.51 (MeO), 26.52 and 24.94 (Me₂C), 24.16 (CH₂-Ph).

Anal. Calc. for C39H39N3O6: C, 72.53; H, 6.08; N, 6.50. Found: C, 72.42; H, 5.83; N, 6.47.

6β: Rf: 0.15 (hexane-ethyl acetate 7:2); m.p. 80 °C; $[\alpha]_D^{20}$ -22.1° (c 1.06, CHCl₃); UV \downarrow_{max} (CHCl₃): 290 nm (ε 272), 260 nm (ε 1450) and 241 nm (ε 6810); IR ν_{max} (KBr): 3080-3020, 2980-2860, 1740, 1720, 1570, 1450, 1370, 1250, 1215, 1075, 860, 760, 740 and 700 cm-1. 1H-N.M.R. (δ): 7.5-7.1 (m, 2 OH, Ph and Tr), 5.7 and 5.53 (2d, 2x1H, 15.6 Hz, CH₂-Ph), 4.48 (dd, 1H, 4.1 and 6.8 Hz, H-3'), 4.35 (dd, 1H, 5.8 and 6.8 Hz, H-2'), 4.15-4.0 (m, 2H, H-1' and H-4'), 3.93 (s, 3H, MeO), 3.41 (dd, 1H, 3.9 and 14.3 Hz, CH₂a-C=), 3.35 (dd, 1H, 3.5 and 10.4 Hz, H-5'a), 3.17 (dd, 1H, 5 and 10.4 Hz, H-5'b), 3.0 (dd, 1H, 7.7 and 14.3 Hz, CH₂b-C=), 1.49 and 1.28 (2s, 6H, Me₂C). 13C-N.M.R. (δ): 161.9 (C=O),

143.7-126.81 (Ph and Tr), 114 (Me₂C), 83.5, 83.26, 82.74 and 81.88 (C-1', C-2', C-3' and C-4'), 63.68 (C-5'), 52.32 (MeO), 51.71 (CH₂-C=), 27.48 and 25.59 (Me₂C), 26.84 (CH₂-Ph).

Anal. Calc. for $C_{39}H_{39}N_{306}$: C, 72.53; H, 6.08; N, 6.50. Found: C, 72.51; H, 6.00; N, 6.43.

Synthesis of 4-methoxy-6-(2,3-0-isopropylidene-5-0trityl- α and β -D-ribofuranosyl) methylpyrimidine (11 $\alpha\beta$) and 4-amino-6-(2,3- $\underline{0}$ -isopropylidene-5- $\underline{0}$ -trityl- α and β -D-ribofuranosyl) methylpyrimidine (12 $\alpha\beta$). A solution of 0.145 g (0.267 mmol) of $10\alpha\beta$ in 30 mL of methanol saturated with ammonia was heated in a sealed tube for 23 h at 90°C. After this time TLC showed the absence of $10\alpha\beta$ and the presence at least two new products. The solvent was evaporated under vacuo and the residue dissolved in a little of ethyl acetate. Filtration and reconcentration of the solution gave a syrup (0.110 g) which was resolved by column chromatography. Elution with hexane-ethyl acetate (2.5:1) gave 0.030 (21%) of $11\alpha\beta$, (α/β , 5:1). Elution with hexane-ethyl acetate (1:10) and then pure ethyl acetate, gave 0.060 g (43%) of $12\alpha\beta$, $(\alpha/\beta$, 5:1).

11 $\alpha\beta$: Rf 0.49 hexane-ethyl acetate (2:1); UV λ_{max} 230 nm; MS m/z 523 (M+ -15), 295 (M+ -Tr), 243 (Tr+).

Anal. Calc. for C33H34N2O5: C, 73.58; H, 6.36; N, 5.20. Found: C, 73.72; H, 6.54; N, 5.02.

12 $\alpha\beta$: Rf 0.54 ethyl acetate; UV $\frac{1}{2}$ max (MeOH) 244 nm (ϵ 6.27x103) and 260 nm (ϵ 3.89x103); MS m/z 281 (M+ -Tr), 265 (M+ -TrO), 243 (Tr+).

Anal. Calc. for C32H33N3O4x1.5 H2O: C, 69.80; H, 6.59; N, 7.63. Found: C, 69.71; H, 6.79; N, 7.26.

Synthesis of $6-(2,3-\underline{0}-isopropylidene-5-\underline{0}-trityl-\alpha$ and $\beta-D-ribofuranosyl)methyl-4-hydroxypyrimidine (13<math>\alpha\beta$). Method \underline{A} . To a solution of 230 mg (0.40 mmol) of $3\alpha\beta$ in 25 mL of ethanol, an excess of Ni-Raney in 5 mL of ethanol was added, besides a few drops of ammonia. The reaction was heated at

reflux for one day, was filtered and the products were separated by column chromatography on silica gel (hexane-ethyl acetate 2:3) to yield 79 mg of 13α and 49 mg of 13β (overall yield 60.7%). UV of the mixture, λ_{max} (MeOH): 233 nm (ϵ 3220); IR vmax (KBr): 3700-2300, 3110, 3090, 3050, 3000, 2975, 2950, 2900, 2000-1750, 1675, 1615, 1550, 1500, 1450, 1420, 1380, 1340, 1270, 1220, 1170, 1150-960, 875, 810, 770, 710 and 630 cm⁻¹; MS m/z: 509 (M+ -15), 494 (M+ -30), 447 (M+ -Ph), 281 (M+ -Tr), 243 (100%). Rf 0.39 (13 α), 0.33 (13 β) (hexane-ethyl acetate 1:3); [α] α +1.54° (c 0.5, methanol) (13 α), +0.66° (c 1.0, methanol) (13 β).

The same procedure was applied to separated epimers 3α and 3β , obtaining the corresponding reduced products 13α and 13β , respectively, without any observed epimerization.

Anal. Calc. for C₃₂H₃₂N₂O₅.2H₂O: C, 68.55; H, 6.47; N, 5.00. Found: C, 69.09; H, 6.67; N, 5.00.

Method B.- A solution of 112 mg (0.206 mmol) of $10\alpha\beta$ in 7 mL of 10% ethanolic KOH was heated at reflux for 15 h. The disappearance of $10\alpha\beta$ was followed with TLC (hexane-ethyl acetate 2:1). After this time, TLC showed the presence of only one new product. The solution was neutralized with concentrated hydrochloric acid, extracted several times with chloroform and the organic layers concentrated giving 99 mg (90%) of $13\alpha\beta$ as a yellow product, $(\alpha/\beta, 2:1)$. The mixture has the same spectra as previously reported 10.

Synthesis of 4-chloro-6-(2,3:5,6-di-0-isopropylidene- α and β -D-mannofuranosyl)methylpyrimidine (14 $\alpha\beta$), and of Z- and E-4-chloro-6-[5-hydroxy-3,4:6,7-bis(isopropylidenedioxy)-1-hepten-1-yl]pyrimidine (14EZ). A solution of 1 g (3.86 mmol) of 2,3:5,6-di-0-isopropylidene-D-mannofuranose (8) and 1.8 g (4.6 mmol) of 4-chloropyrimidin-6-yl-methylenetriphenyl-phosphorane (9) in 6 mL of methylene chloride was heated at reflux for 36 h. The solvent was eliminated and the residue was dissolved in methanol, filtered, and then purified by column chromatography (hexane-ethyl acetate 83:17), yield-

ing 1.07 g (75%) of a mixture of 14E, 14Z and 14 β (in a 1:4:16 ratio) as a transparent pale violet oil and 143 mg (10%) of 14 α as a yellow syrup. The h irradiation of the mixture results in a rapid stereoselective cyclization of the open chain isomer 14EZ to the cyclic 14 β .

14 $\alpha\beta$: UV max (MeOH): λ 248 nm (ϵ 3560) and 210 nm (ϵ 3000); IR ν_{max} (KBr): 3000, 2960, 2910, 2875, 1575, 1540, 1470, 1380, 1325, 1275, 1215, 1175, 1130, 1110, 1100, 1070, 1000, 980, 930, 900, 870, 850 and 750 cm⁻¹; MS m/z: 373 and 371 (M+ +1), 357 and 355 (M+ -Me), 314 and 312 (M+ -Me₂CO), 299 and 297 (M+ -Me₂CO -Me). Rf: 0.40 (14 α), 0.48 (14 β) (hexane-ethyl acetate 2:1); $[\alpha]_{D}^{20} = +54^{\circ}$ (14 α), -17.5° (14 β) (c 1, MeOH).

14E: 1H-NMR (from the mixture) (δ): 8.95 (d, 1H, 1 Hz, H-2), 7.32 (d, 1H, 1 Hz, H-5), 6.67 (d, 1H, 18 Hz, H-1'), 7.25 (dd, 1H, 18 and 8 Hz, H-2'), 4.95 (m, 1H, H-3').

142: 1H-NMR (from the mixture) (δ): 9.1 (s, 1H, H-2), 7.6 (s, 1H, H-5), 6.4 (m, 2H, H-1' and 2'), 5.8 (dd, 1H, 6 and 8 Hz, H-3').

14 $\alpha\beta$ + 14EZ mixture: Anal. Calc. for C17H23ClN2O5: C, 55.06; H, 6.25; N, 7.56. Found: C, 54.69; H, 6.21; N, 7.54.

Synthesis of 4-methoxy-6-(2,3:5,6-di-Q-isopropylidene- α and β -D-mannofuranosyl)methylpyrimidine (15 $\alpha\beta$) and 4-amino- $6-(2,3:5,6-di-\underline{0}-isopropylidene-\alpha$ and $\beta-D-mannofuranosyl) me$ thylpyrimidine (16 $\alpha\beta$). A solution of 0.418 g (1.127 mmol) of $14\alpha\beta$ in 20 mL of methanol saturated with ammonia was heated in a sealed tube for 23 h at 90°C. After this time TLC (AcOEt) showed the absence of $14\alpha\beta$ (Rf 0.79) presence of at least two new products (Rf 0.68 and 0.15). The solvent was evaporated under vacuo and the residue dissolved in a little of ethyl acetate. Filtration and reof the solution gave a syrup (0.400 g) which concentration was resolved by column chromatography. Elution with hexane-ethyl acetate (1:10) gave 0.120 g (29%) of $15\alpha\beta$, (α/β , 1:9) as a colorless oil. Elution with methanol-ethyl acetate (1:5) gave 0.250 g (63.2%) of $16\alpha\beta$, (α/β , 1:6). Epimers 15α and 15β were separated by thick layer chromatography (ethyl acetate:hexane 2:1).

15αβ: Rf 0.68 ethyl acetate; UV λ_{max} (MeOH) 233 nm (ε 1.05x103); IR ν_{max} (film): 3000, 2950, 2890, 1600, 1550, 1475, 1375, 1260, 1210, 1160, 1125, 1070, 1040, 990, 870, 850, 810; MS m/z: 366 (M+), 351 (M+ -Me), 308 (M+ -Me₂CO), 293 (M+ -Me₂CO -Me), 250 (M+ -2Me₂CO).

Anal. Calc. for C₁₈H₂6N₂O₆.2H₂O: C, 56.23; H, 7.34; N, 7.29. Found: C, 56.51; H, 6.90; N, 7.07.

16\alpha\beta: Rf 0.73 ethyl acetate; UV λ_{max} (MeOH) 267 nm (ϵ 3.33×103), 232 nm (ϵ 6.27×103) and 208 nm (ϵ 3.33×103); IR Vmax(film): 3360, 3200, 3000, 2950, 2910, 2875, 1660, 1610, 1540, 1500, 1450, 1385, 1350, 1275, 1220, 1170, 1075, 1000, 925, 900, 850 cm⁻¹; MS m/z 351 (M+), 336 (M+ -Me), 293 (M+ -Me₂CO), 278 (M+ - Me₂CO -Me).

Anal. Calc. for C17H25N3O5.1.5 H2O: C, 53.95; H, 7.46; N, 11.11. Found: C, 53.99; H, 7.59; N, 10.71.

synthesis of $6-(2,3:5,6-di-O-isopropylidene-\beta-D-manno-furanosyl)$ methyl-4-hydroxypyrimidine (17). Method A.- To a solution of 130 mg (0.32mmol) of 4β in 25 mL of ethanol, a two fold excess of Ni-Raney in 5 mL of ethanol was added, besides a few drops of ammonia. After being heated at reflux for one day, the mixture was filtered and the solvent evaporated under vacuo. The residue was purified by column cromatography (ethyl acetate) to yield 97.5 mg of 17β (84.5%) UV of the mixture, λ_{max} (MeOH): 232 nm (ϵ 3220) and 254 nm (ϵ 1930), IR ν_{max} (KBr): 3450, 3300-2750, 3200, 3150, 31000, 3000, 2950, 2920, 1670, 1600, 1540, 1470, 1420, 1390, 1275, 1215, 1175, 1075, 1100, 995, 950, 860, 825 and 810 cm-1. M.S. m/z: 352(M+), 337(M+-15), 294 (M+-Me2CO), 279 (M+-Me2CO -Me) or (M+-Me-HOCN), 252 (M+-Me2CO -Me-CNH).

Anal. Calc. for C17H24N2O6x0.5 H2O: C, 56.50; H, 6.97. Found: C, 56.97; H, 6.80.

Method B.- A solution of 147 mg (0.40 mmol) of $14\alpha\beta$ in 7 mL of 10% of ethanolic KOH was heated at reflux for 14 h. The disappearance of $14\alpha\beta$ was followed with TLC (hexane-ethyl acetate 2:1). After this time, TLC showed the presence of only one new product. The solution was neutralized with concentrated hydrochloric acid, extracted several times with chloroform and the organic layers concentrated giving 125 mg (90%) of $17\alpha\beta$ as a yellow product, $(\alpha/\beta, 2:1)$. The mixture has the same spectra as the ones reported above. Rf: 0.15 (17α) and 0.25 (17β) .

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